

## 9-(Bromoacetyl)anthracene dimer

Kanji Kubo,<sup>a\*</sup> Taisuke Matsumoto,<sup>b</sup> Keiko Ideta,<sup>b</sup> Kenichiro Watanabe<sup>c</sup> and Tadamitsu Sakurai<sup>c</sup>

<sup>a</sup>School of Dentistry, Health Sciences University of Hokkaido, 1757 Kanazawa, Ishikari-Tobetsu, Hokkaido 061-0293, Japan, <sup>b</sup>Institute for Materials Chemistry and Engineering, Kyushu University, Kasuga-koen, Kasuga, Fukuoka 816-8580, Japan, and <sup>c</sup>Department of Materials and Life Chemistry, Faculty of Engineering, Kanagawa University, Kanagawa-ku, Yokohama 221-8686, Japan

Correspondence e-mail:  
kubo-k@hoku-iryu-u.ac.jp

### Key indicators

Single-crystal X-ray study  
T = 153 K  
Mean  $\sigma(C-C)$  = 0.003 Å  
R factor = 0.036  
wR factor = 0.074  
Data-to-parameter ratio = 16.9

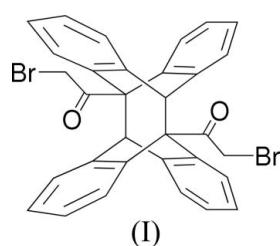
For details of how these key indicators were automatically derived from the article, see  
<http://journals.iucr.org/e>.

The title compound [systematic name: 5,6,11,12-tetrahydro-5,11-bis(bromoacetyl)-5,12[1',2']:6,11[1'',2'']dibenzenodibenz-[a,e]cyclooctene], C<sub>32</sub>H<sub>22</sub>Br<sub>2</sub>O<sub>2</sub>, is a [4+4] photodimer of 9-(bromoacetyl)anthracene. The molecule is centrosymmetric and contains a long Csp<sup>3</sup>–Csp<sup>3</sup> bond of 1.619 (3) Å. Intermolecular C–H···O interactions are observed.

Received 15 February 2007  
Accepted 18 April 2007

### Comment

Anthracene and its derivatives have been studied extensively in the fields of photochemistry and photobiology. Photodimerization of anthracene is one of the oldest known photochemical reactions and a classical example of photocycloaddition (Becker, 1993; Bouas-Laurent *et al.*, 2000). Recently, the photochemical reaction (Matsumoto *et al.*, 1974), laser flash photolysis (Suzuki *et al.*, 2004) and crystal structure (Kubo *et al.*, 2007) of 9-(bromoacetyl)anthracene have been reported, but its photodimer has not been obtained. The title compound, (I), was obtained unexpectedly when growing single crystals of 9-(bromoacetyl)anthracene from a chloroform solution under sunlight. We now report the structure of (I), with the aim of contributing to a deeper understanding of the [4+4] photodimerization of 9-(bromoacetyl)anthracene.



Compound (I) is centrosymmetric and a highly strained butterfly-shaped anthracene dimer (Fig. 1). The angle of the intersection between the least-squares plane defined by C1–C6/C13/C14 and the plane defined by C6–C13 is 43.1 (1)°, which is similar to that of the intramolecular photocyclomer of 9-anthroic anhydride [47.6 (1)°; Cicogna *et al.*, 2002]. The two bromoacetyl groups are *anti* with respect to one another across the anthracene dimer. Compound (I) contains a long central C–C bond connecting the anthracene units, compared with the standard C–C single-bond length of 1.54 Å (Battersby *et al.*, 1995). The C–C bond length [1.619 (3) Å] is close to that of the anthracene dimer [1.618 (3) Å; Abboud *et al.*, 1990].

Intermolecular C–H···O interactions with typical geometric parameters (Takahashi *et al.*, 2005; Kubo *et al.*, 1997, 2006, 2007) are observed in the crystal structure (Table 1).

## Experimental

Compound (I) was prepared by sunlight irradiation of a chloroform solution ( $1.25 \text{ mmol l}^{-1}$ ) of 9-(bromoacetyl)anthracene. Crystals of (I) were grown from a chloroform solution by slow evaporation.

### Crystal data

$C_{32}H_{22}Br_2O_2$	$\gamma = 83.656 (11)^\circ$
$M_r = 598.33$	$V = 608.0 (2) \text{ \AA}^3$
Triclinic, $P\bar{1}$	$Z = 1$
$a = 7.9044 (18) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 9.1861 (17) \text{ \AA}$	$\mu = 3.37 \text{ mm}^{-1}$
$c = 9.4138 (17) \text{ \AA}$	$T = 153.1 \text{ K}$
$\alpha = 66.648 (9)^\circ$	$0.18 \times 0.08 \times 0.07 \text{ mm}$
$\beta = 75.696 (10)^\circ$	

### Data collection

Rigaku Saturn diffractometer  
Absorption correction: multi-scan  
(REQAB; Jacobson, 1998)  
 $R_{\min} = 0.634$ ,  $T_{\max} = 0.790$

7481 measured reflections  
2777 independent reflections  
2444 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.033$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.036$   
 $wR(F^2) = 0.074$   
 $S = 1.05$   
2777 reflections

164 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.56 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.48 \text{ e \AA}^{-3}$

**Table 1**  
Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C9-H7\cdots O1^{ii}$	0.95	2.59	3.318 (4)	134

Symmetry code: (ii)  $x, y - 1, z$ .

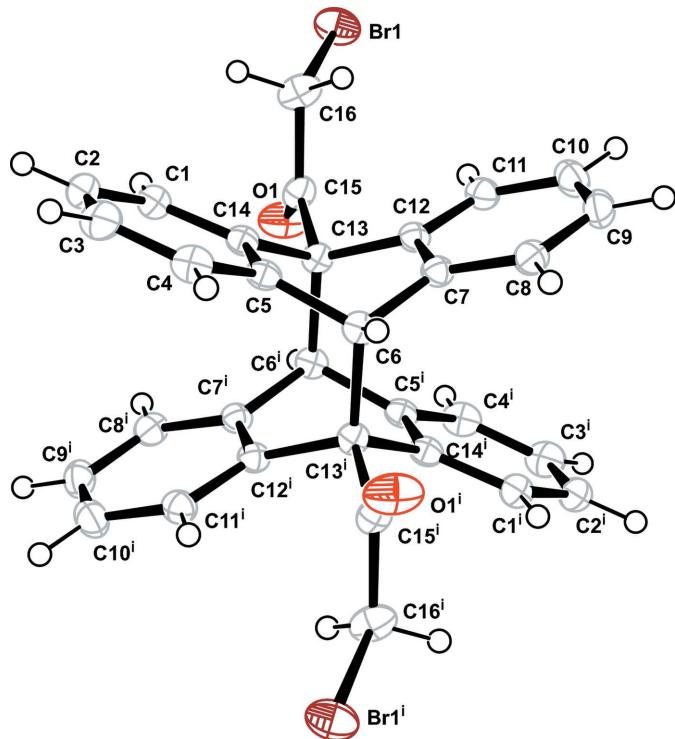
H atoms were positioned geometrically and refined as riding, with  $C-H = 0.95 \text{ \AA}$  for aromatic,  $C-H = 0.99 \text{ \AA}$  for methylene and  $C-H = 1.00 \text{ \AA}$  for methine H atoms, and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

Data collection: *CrystalClear* (Rigaku, 1999); cell refinement: *CrystalClear*; data reduction: *CrystalStructure* (Rigaku/MSC, 2006); program(s) used to solve structure: *SIR97* (Altomare *et al.*, 1999); program(s) used to refine structure: *SHELXL97* (Sheldrick 1997); molecular graphics: *ORTEPIII* (Burnett & Johnson, 1996) and *Mercury* (Version 1.3; Bruno *et al.*, 2002); software used to prepare material for publication: *CrystalStructure*.

This research was partially supported by a Frontier Research Project from the Ministry of Education, Sports, Culture, Science and Technology, Japan.

## References

- Abboud, K. A., Simonsen, S. H. & Roberts, R. M. (1990). *Acta Cryst.* **C46**, 2494–2496.



**Figure 1**

The molecular structure of (I), showing the atom-numbering scheme and 50% probability displacement ellipsoids. [Symmetry code: (i)  $1 - x, -y, 2 - z$ .]

- Altomare, A., Burla, M. C., Camalli, M., Cascarano, G. L., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). *J. Appl. Cryst.* **32**, 115–119.
- Battersby, T., Gantzel, P., Baldridge, K. & Siegel, J. (1995). *Tetrahedron Lett.* **36**, 845–848.
- Becker, H. D. (1993). *Chem. Rev.* **93**, 145–172.
- Bouas-Laurent, H., Castellan, A., Desvergne, J.-P. & Lapouyade, R. (2000). *Chem. Soc. Rev.* **29**, 43–55.
- Bruno, I. J., Cole, J. C., Edgington, P. R., Kessler, M., Macrae, C. F., McCabe, P., Pearson, J. & Taylor, R. (2002). *Acta Cryst.* **B58**, 389–397.
- Burnett, M. N. & Johnson, C. K. (1996). *ORTEPIII*. Report ORNL-6895. Oak Ridge National Laboratory, Tennessee, USA.
- Cicogna, F., Ingrosso, G. & Marchetti, F. (2002). *Acta Cryst.* **C58**, o359–o361.
- Jacobson, R. (1998). *REQAB*. Private communication to Rigaku Corporation, Tokyo, Japan.
- Kubo, K., Fukeda, E., Matsumoto, T., Endo, K. & Mori, A. (2006). *Acta Cryst.* **E62**, o2986–o2987.
- Kubo, K., Kato, N. & Sakurai, T. (1997). *Acta Cryst.* **C53**, 132–134.
- Kubo, K., Watanabe, K. & Sakurai, T. (2007). *Acta Cryst.* **E63**, o1300–o1301.
- Matsumoto, T., Sato, M. & Hirayama, S. (1974). *Bull. Chem. Soc. Jpn.* **47**, 358–363.
- Rigaku (1999). *CrystalClear*. Rigaku Corporation, Tokyo, Japan.
- Rigaku/MSC (2006). *CrystalStructure*. Version 3.8. Rigaku/MSC, The Woodlands, Texas, USA.
- Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.
- Suzuki, T., Kaneko, Y., Ikegami, M. & Arai, T. (2004). *Bull. Chem. Soc. Jpn.* **77**, 801–806.
- Takahashi, H., Takechi, H., Kubo, K. & Matsumoto, T. (2005). *Acta Cryst.* **E61**, o3041–o3043.